

Anesthetics

Lecture 12 part 2





Inhalational Anesthetics

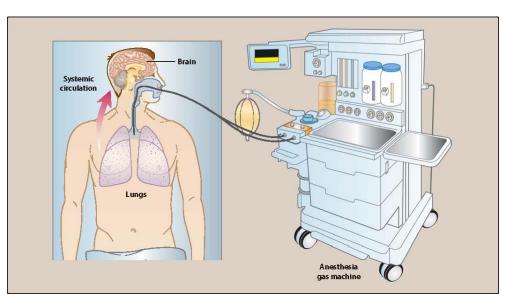




Inhalational Anesthetics

- Primarily used for <u>maintenance</u> of anesthesia following induction by IV agents.
- Depth of anesthesia correlates with inhaled concentration.
- Less risk of cardiac/respiratory depression than IV agents.
- No antagonists.

usually termination the effect by redistribution

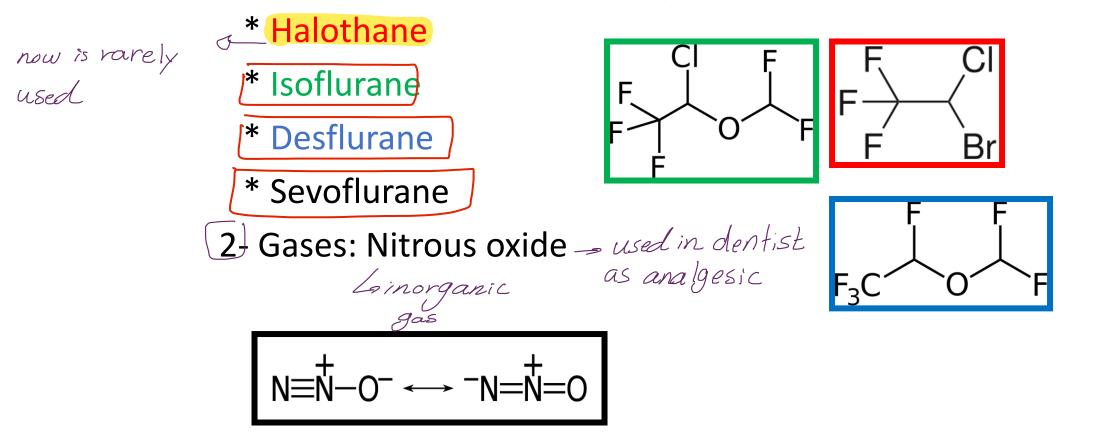






Inhaled anesthetics

1- Halogenated(with Cl⁻, F⁻, I⁻) Volatile liquids:







Mechanism of Action of Inhalational Anesthetics is UNKNOWN!

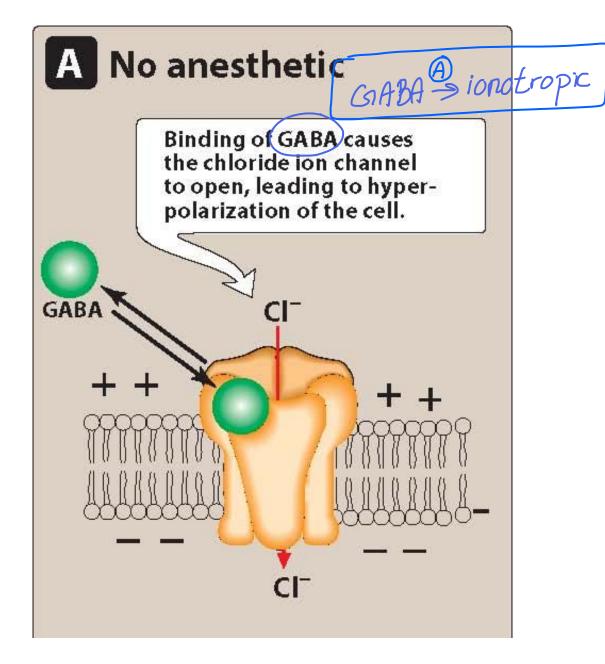
Possible mechanisms:

Increase the sensitivity of GABA_A receptors to GABA

(nitrous oxide, ketamine have no effect on GABA)

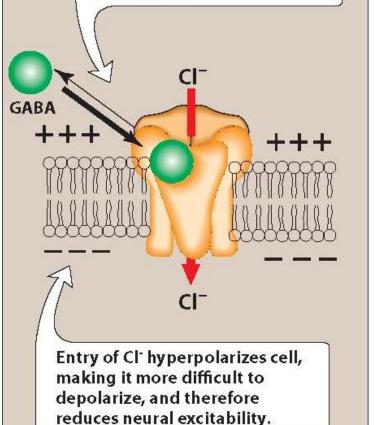
- Increase the activity of glycine receptors in the spinal chord
- Block excitatory postsynaptic currents of nicotinic receptors





B In presence of inhaled anesthetic

Binding of GABA is enhanced by inhaled anesthetics, resulting in a greater entry of chloride ion.



like Bz and barbiturate



to induce anesthisia in e uppel inhalation anesthetic is to concentration of PE 50% of PE Potency: MAC

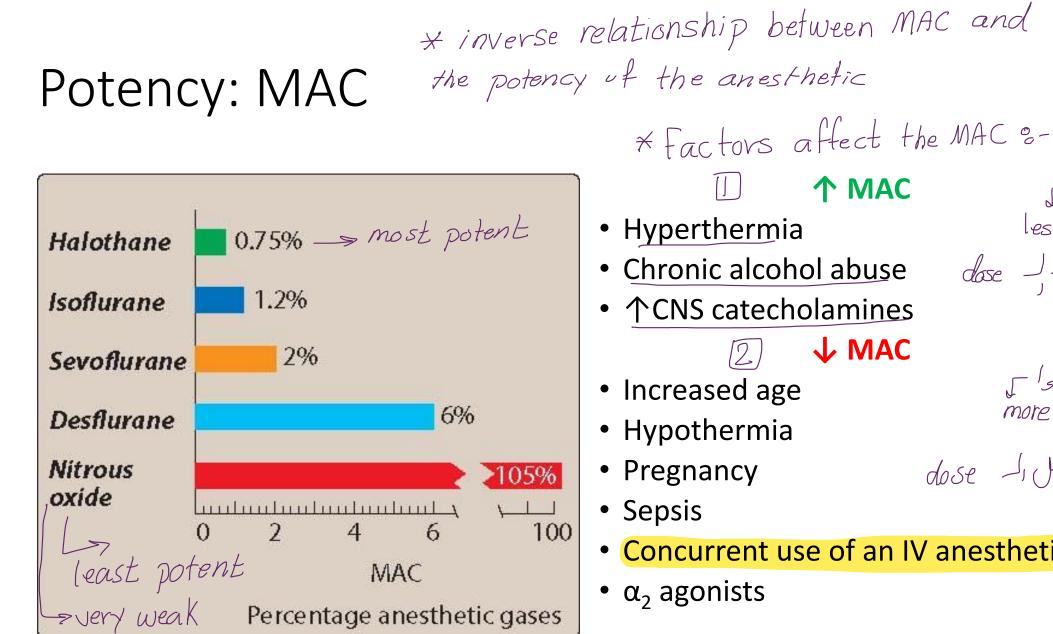


Finishing of the tidal cycle of breathing

Minimum Alveolar Concentration (MAC)

- The end-tidal concentration of an inhalational anesthetic needed to eliminate movement in 50% of patients stimulated by a standardized – incision. Incision. the pt completely unconsciousness, loosing reflexes and motor function.
- MAC = \underline{ED}_{50} of an anesthetic
- MAC is expressed as percentage of alveolar gas mixture/ partial wound pressure as % of 760 mm of Hg.







× خلوالدو ک

less potent dose _ in the x Chronic alcohol abuse 个CNS catecholamines **↓** MAC

个 MAC

- Hypothermia
- Pregnancy
- Sepsis
- Concurrent use of an IV anesthetic
- α_2 agonists

dose Ji ter rilis x





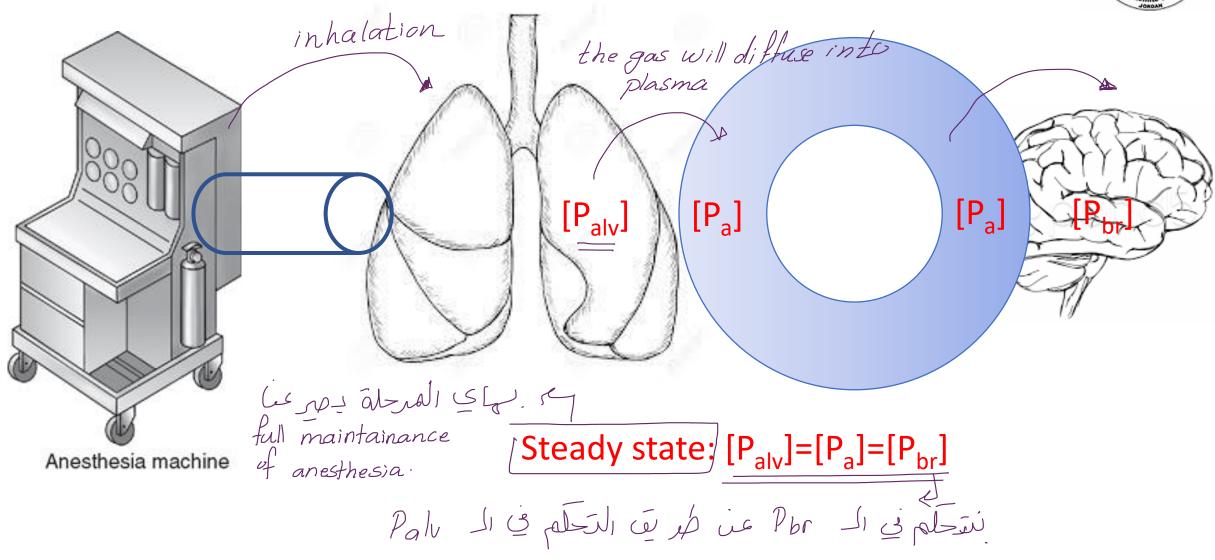
Distribution

The pharmacologic effect of an inhalation agent is determined by the partial pressure of the anesthetic in the brain $[P_{br}]$

 $[P_{br}]$ depends on alveolar partial pressure $[P_{alv}]$ which is controlled by pressure at the origin of the respiratory pathway.







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Factors affecting equilibrium/steady state

I. Alveolar Wash-In

"Replacement of normal lung gases with inspired anesthetic mixture"

- II. Anesthetic Uptake "most important
- a. Solubility in blood
- b. Cardiac output
- c. Tissue type
- d. Alveolar:venous gradient



Solubility

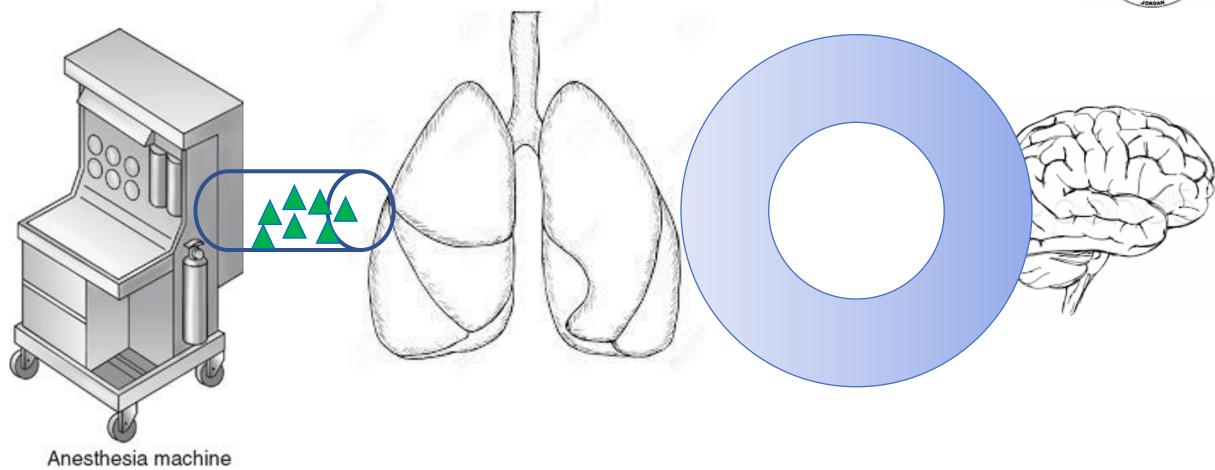
[anesthetic] in blood = coefficient [anesthetic] in a leveoli _ oute / Soluble in the blood 'en, so in some



- Determined by blood:gas partition coefficient [the ratio of the concentration of the anesthetic in the blood to the concentration of the anesthetic in the gas phase=solubility of an anesthetic in blood (aleveoli) pitblood: gas coefficient is Low
- Low blood solubility \rightarrow few anesthetic molecules are required to raise $[P_a] \rightarrow Less$ time for induction and recovery rapid
- High blood solubility → more anesthetic molecules are required to
 raise $[P_a] \rightarrow more$ time for induction and recovery

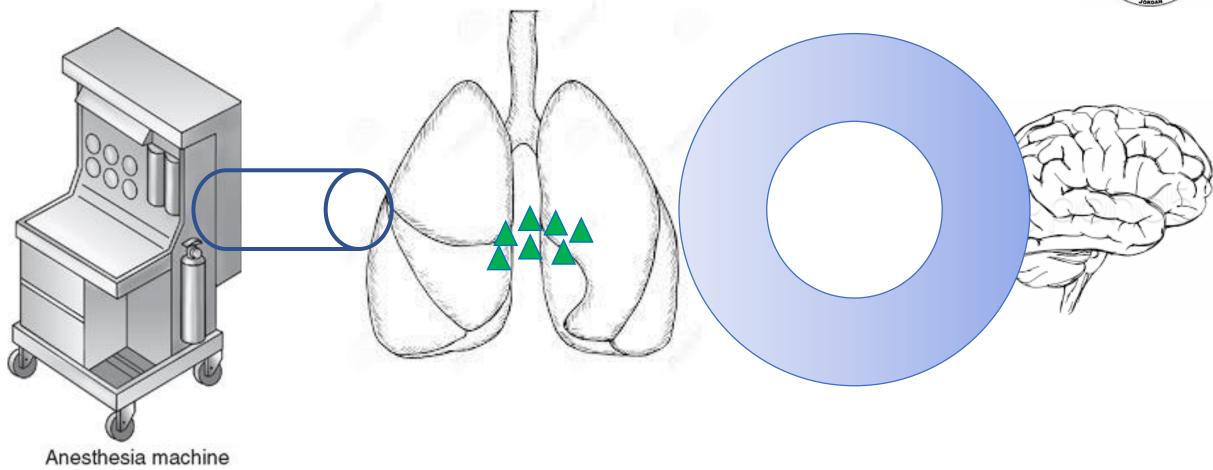
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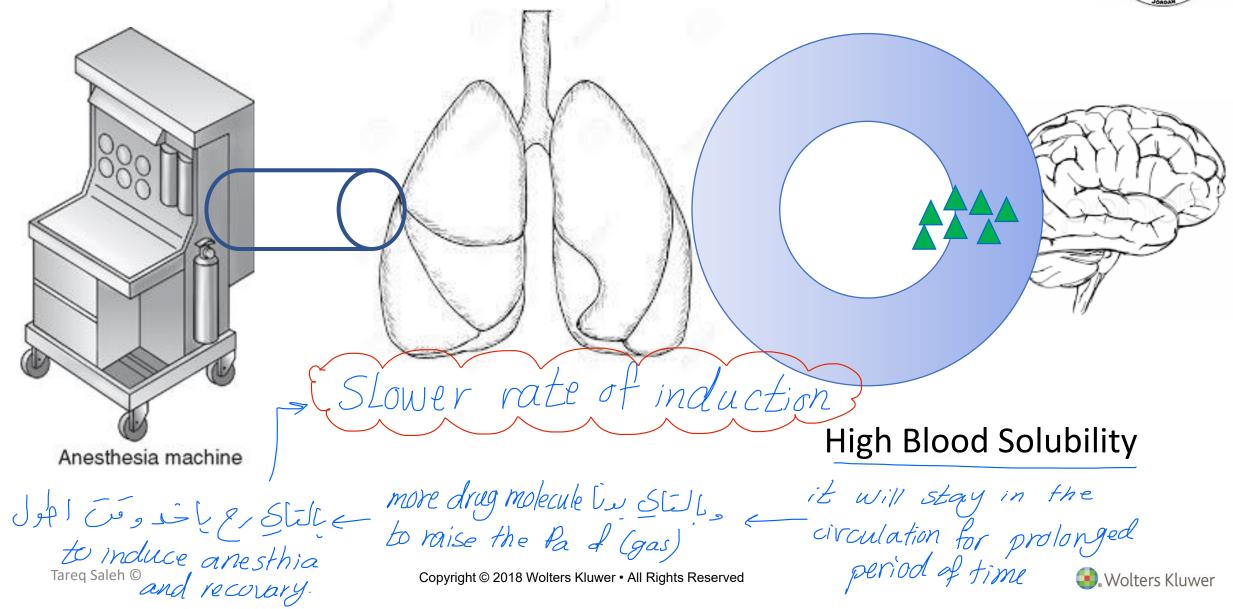




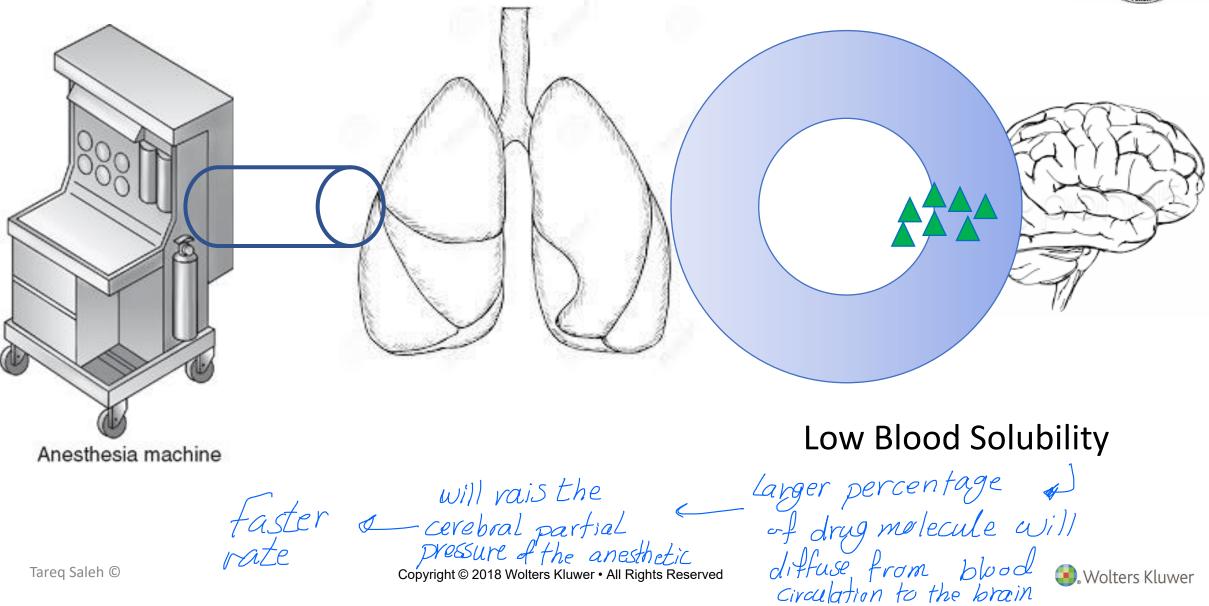






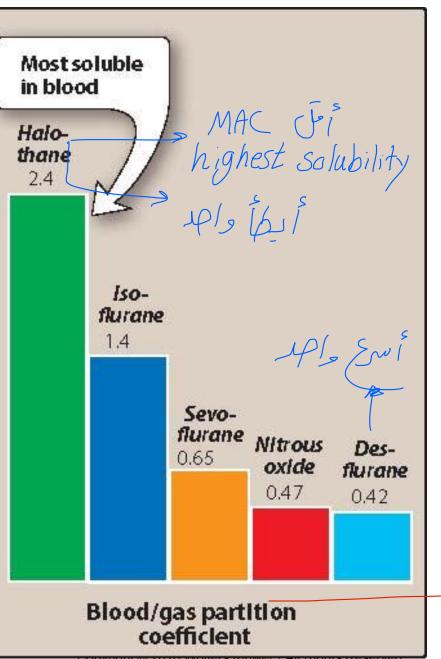






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inverse relationship D MAC (JMAC-> the drug is more potent) 2 solubility (blood: gas coefficient -> the drug is faster in induction anesthsia)





Steady Neger X X state

Trui

the Low solubility

98 لأن لو جان الرسمة في سارير نعدر تحد مس ال MAC فقط من الرفوك .. > most potent one ? dependion MAC only * We cant detrmine in Molters Kluwer

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* invers relation ship

Cardiac Output



es il vi els , L provery J vi is • CO affects washing the anesthetic to peripheral tissue (NOT the site of action) High CO induction brain. Is is ins **ANESTHETICS?**

Low CO $\rightarrow \downarrow$ pulmonary blood flow (less removal to the periphery) \rightarrow **fast** rise in $[P_{alv}] \rightarrow$ **faster** induction





Differences in Tissue Type on Uptake

Steady State '	~	Blood flow to the tissue				
Steady State	•	Capacity of tissue to store the anesthetic (proportional to tissue volume)				
Tis	sue Type	Perfusion (Blood Flow)	Capacity			
Brain, heart, liver , kidney, endocrine glands		Good	Low			
Skele	tal muscles	Poor	Large			
Adip	oose tissue	Poor	Large			
Tareq Saleh © Bon	e, cartilage ^{copyright (}	© 2018 Wolters Kluv Poör ghts Reserved	Low			

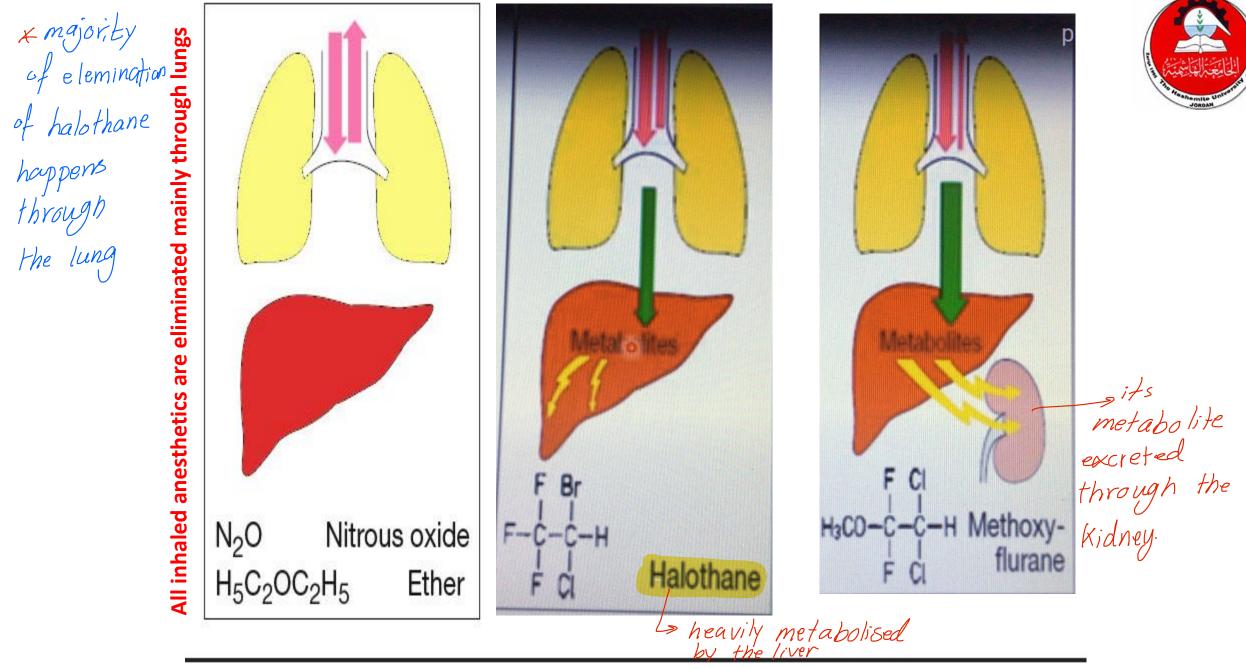




Elimination

- The time to recovery from inhalation anesthesia depends on the rate of elimination of anesthetics from the brain after the inspired concentration of anesthetic has been decreased.
- Inhaled anesthetics that are relatively insoluble in blood (low blood: gas partition coefficient) and brain are eliminated at faster rates than more soluble anesthetics.





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Elimination routes of different volatile anesthetics

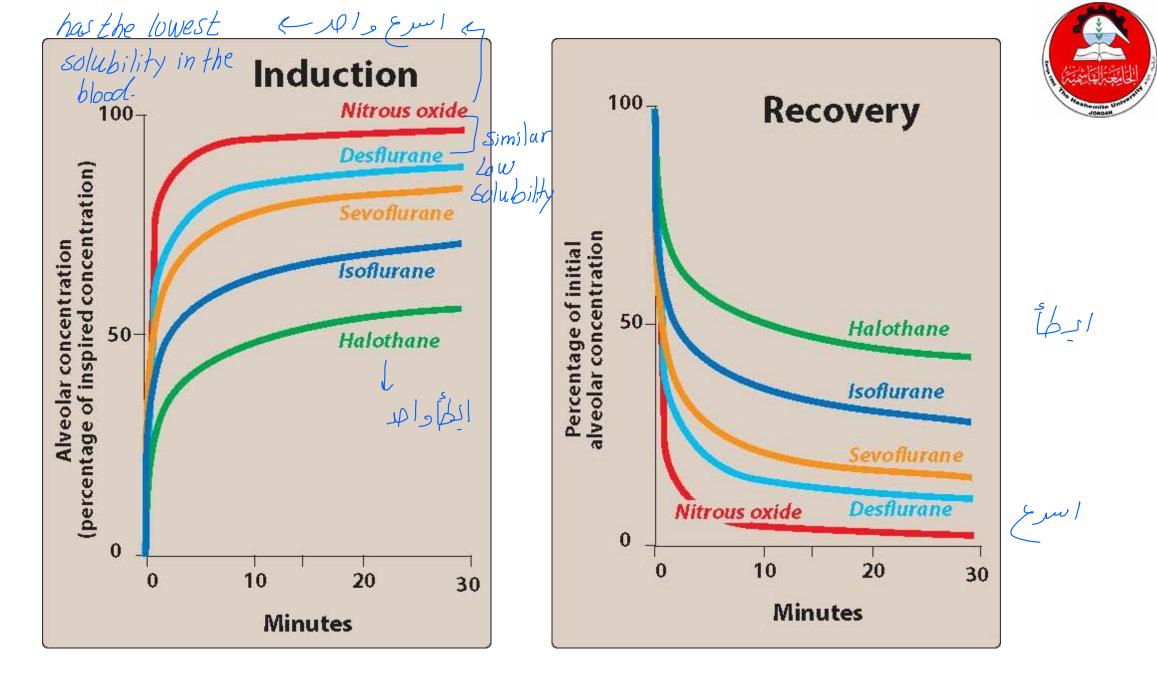




Recovery

- The duration of exposure to the anesthetic can have a marked effect on the time of recovery. If exposure to the anesthetic is short, recovery may be rapid.
- Clearance of inhaled anesthetics by the <u>lungs into the expired</u> air is the major route of their elimination from the body











- Has a pungent smell \rightarrow stimulates the respiratory reflexes \rightarrow NOT used for inhalational induction ONLY for maintainance.
- Causes hypotension
- Solubility? Induction time?
- Low cost
- Longer surgeries





Desflurane

ONLY For maintainance as isoflurane.

- Respiratory irritant \rightarrow NOT used for inhalational induction
- Causes hypotention
- Low blood solubility --- very fast induction and recovery
- Higher cost
- Better for short surgeries

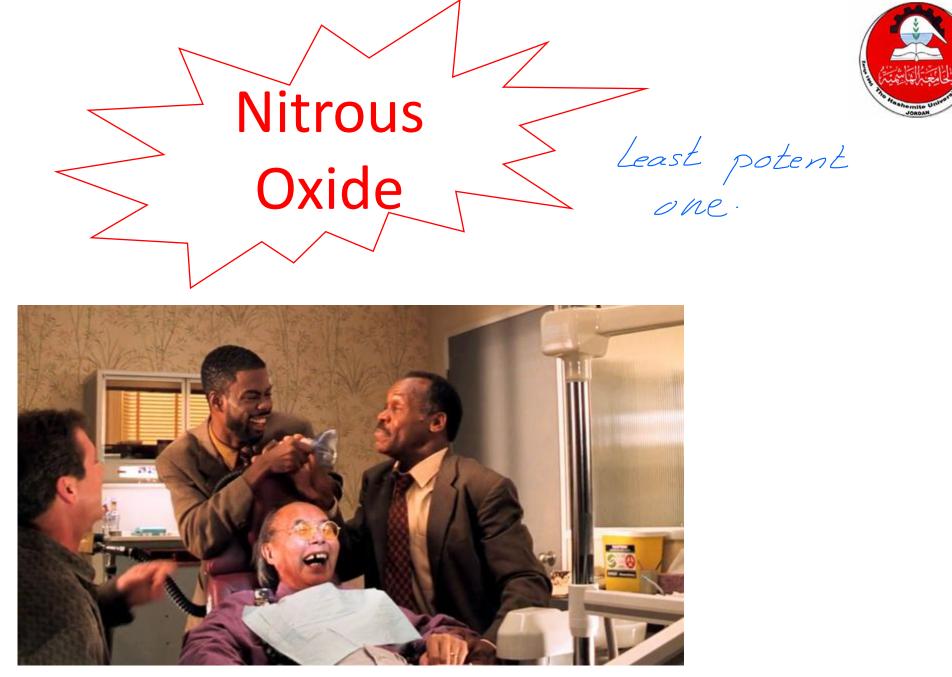




Sevoflurane ---> Used to induce a nesthesia in pediatric age population

- Low pungency and respiratory irritation → <u>can be used for</u> <u>inhalational induction</u>
- Low solubility









Nitrous Oxide

- <u>Gas</u> (inorganic).
- Very rapid induction and recovery. - Why? very low blood solubility.
- least potent, highest MAC value.
- Poor anesthetic, good analgesic
- Administered with O₂ to avoid diffusion hypoxia (to produce sedation dentistry)
- Administered with other inhalational agents for general anesthesia





	Halothane	lsoflurane	Desflurane	Sevoflurane
Arrhythmias	Increased			
Dopamine + Norepinephrine + Epinephrine Sensitivity to catecholamines	Increased			—
Cardiac output	Decreased	Decreased to a lesser extent than <i>halothane</i>	Decreased to a lesser extent than <i>halothane</i>	Decreased to a lesser extent than halothane
Blood pressure	Dose dependent decreased	Dose dependent decreased	Dose dependent decreased	Dose dependent decreased
Respiratory reflexes	Inhibited	Initial stimulation	Initial stimulation	Inhibited
Hepatic toxicity	Some	Low risk	Low risk	Low risk
Renal toxicity	Low risk	Low risk	Low risk	Some risk

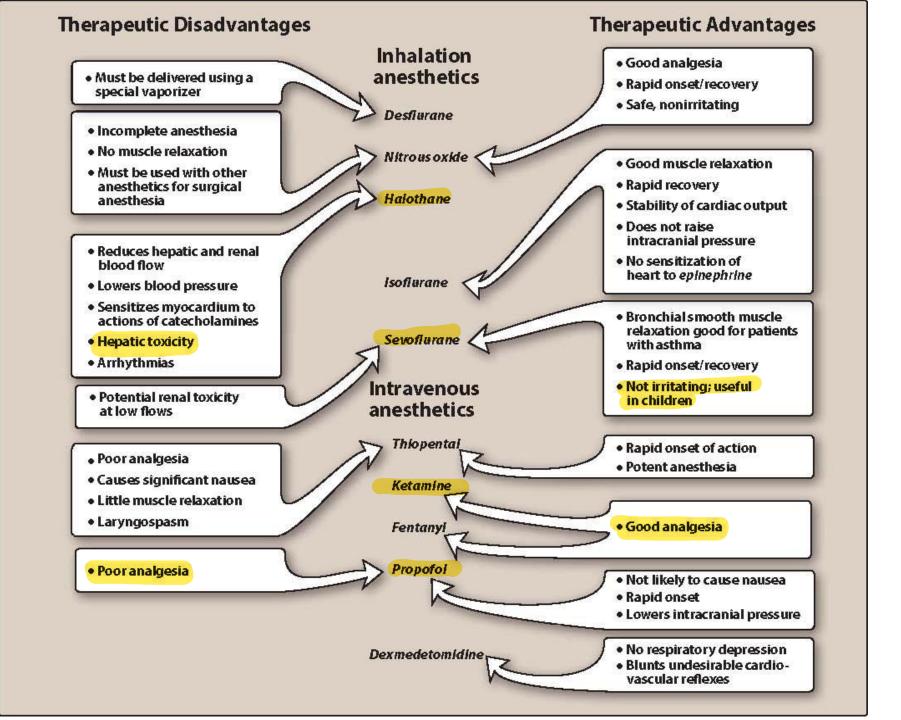




Malignant Hyperthermia

- Rare anesthesia complication (only in susceptible patients; autosomal dominant)
- Exposure to: halogenated anesthetics, succinylcholine
- Life threatening
- Due to uncontrolled, excessive increase in skeletal muscle oxidative metabolism
- Treatment: dantrolene





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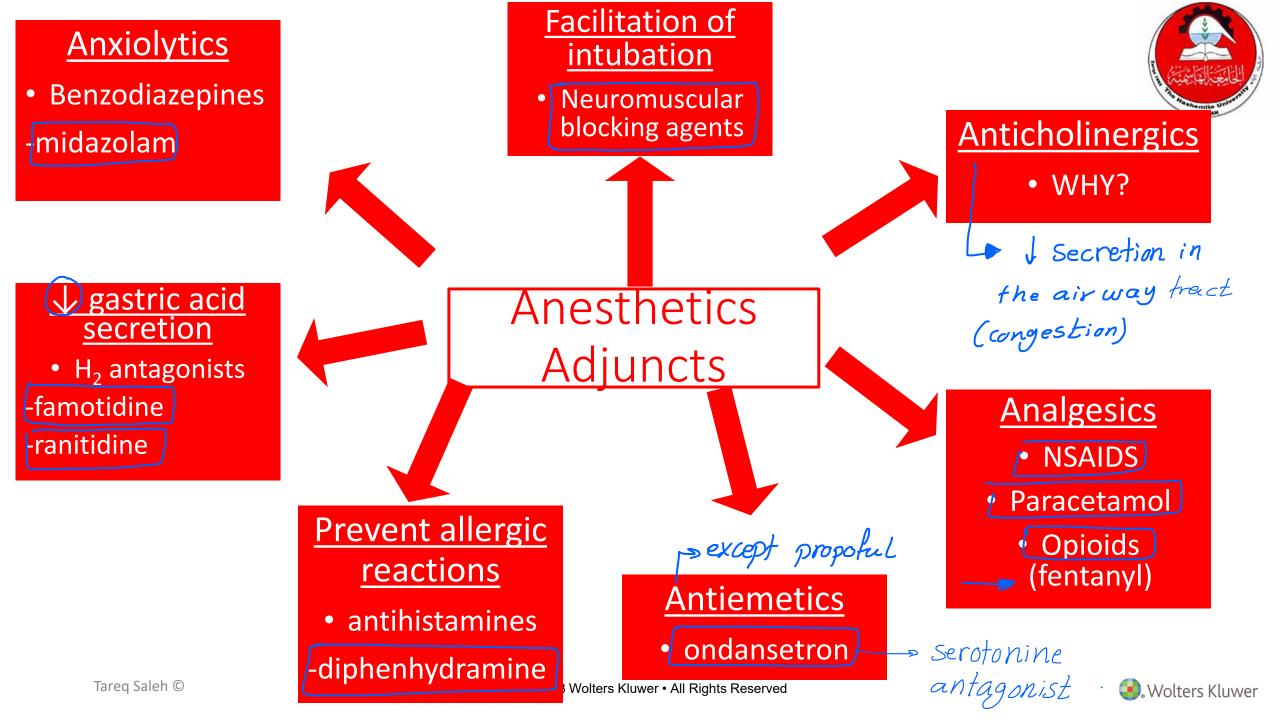
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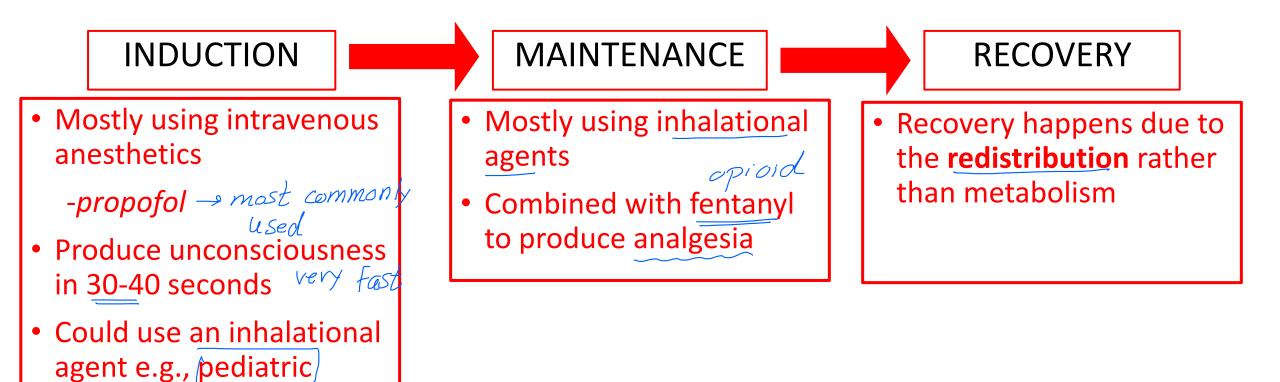
Anesthetic Adjuncts







Stages of Anesthesia



La like sevoflurane





Local Anesthetics



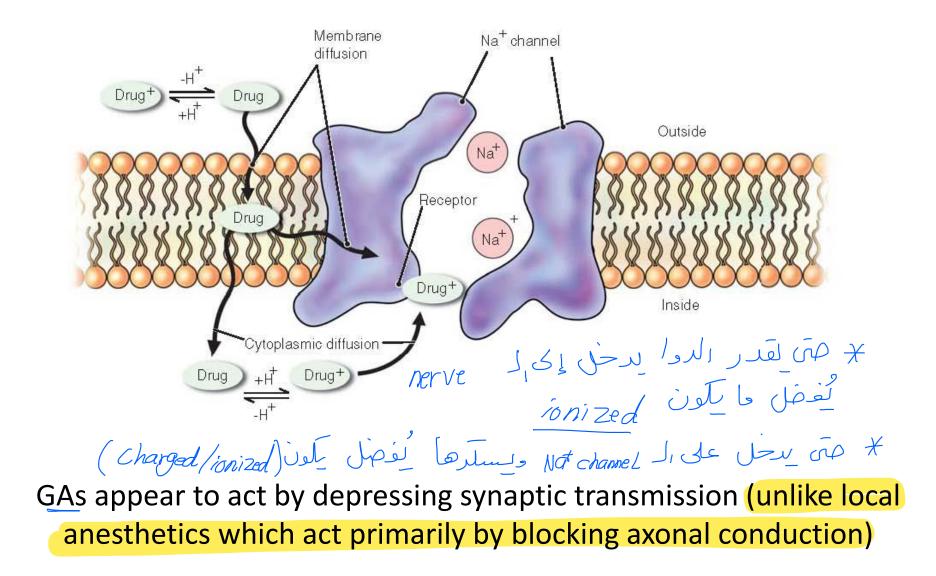


Local Anesthetics

- Low doses: block sensory conduction
- High doses: block motor impulses (paralysis)
- Mechanism of action: "Sodium channels blockade"









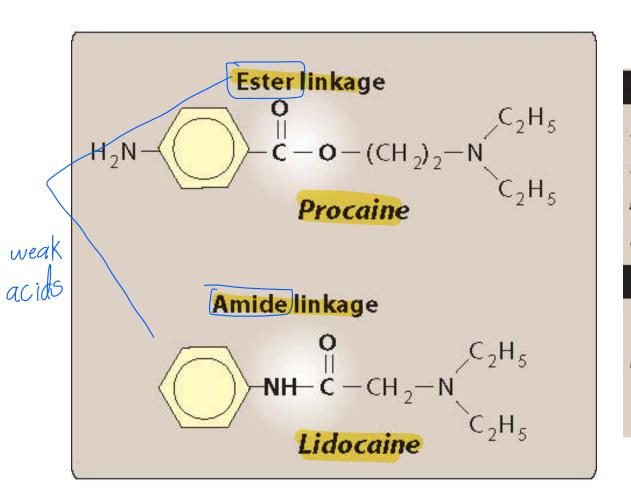


Delivery Options

- Topical
- Infiltration (dentist)
- Perineural
- Neuraxial
 - Spinal
 - Epidural
 - Caudal







LOCAL ANESTHETICS: AMIDES

Bupivacaine MARCAINE Lidocaine XYLOCAINE Mepivacaine CARBOCAINE Ropivacaine NAROPIN

LOCAL ANESTHETICS: ESTERS

Chloroprocaine NESACAINE Procaine NOVOCAINE Tetracaine PONTOCAINE





Local Anesthetics

Actions:

- Vasodilation) + sensory block
 - leads to rapid diffusion \rightarrow short duration of action
 - overcome by adding a vasoconstrictor e.g., *epinephrine*

to increse the duration of action

Antiarrhythmic

- e.g., *lidocaine*



Local Anesthetics



Duration of actions:

- Factors affecting the duration of action:
- **1.** Tissue pH
- 2. Nerve morphology my linated
 3. Concentration
- 3. Concentration

- Hepatic metabolism does NOT affect duration of action of local anesthetics
- - lower pKa \rightarrow more ionized at physiologic pH \rightarrow faster
 - What happens if the tissue is infected? L> pH-> acidic -> less drug available in ionized form.

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Systemic Toxicity

- What if a local anesthetic was administered frequently or inadvertently in the vein (IV)?
- □ Local Anesthetic Systemic Toxicity (LAST)
- 1. Altered mental status
- 2. Seizures
- 3. Cardiovascular instability

Treatment: Lipid Rescue Therapy (20% lipid emulsion infusion)





CHARACTERIST	пс	• Procaine ESTERS • Chloroprocaine	•Tetracaine •Cocaine	AMIDE	S + Lidocaine + Mepivacaine S + Bupivacaine + Prilocaine + Ropivacaine		
Metabolism		Rapid by plasma cholinesterase		Slow, hepatic			
Systemic toxicity		Less likely		More likely			
Allergic reaction		Possible- PABA derivatives form		Very	Very rare		
Stability in solution	on	Breaks down in ampules (heat, sun)		Very stable chemically			
Onset of action		Slow as a general rule		Moderate to fast			
pKa's	Higher than physiologic ph		logic pH (8.5–8.9)	Close to physiologic pH (7.6–8.1)			
DRUG		POTENCY	ONSET		DURATION		
Procaine		Low	Rapid		Short		
Chloroprocaine		Low	Rapid		Short		
Tetracaine		High	Slow		Long (spinal)		
Lidocaine	Low		Rapid		Intermediate		
Mepivacaine	Low		Moderate		Intermediate		
Bupivacaine		High	Slow		Long		
Ropivacaine	High		Moderate		Long		





- Thank you
- Questions?

